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Investigations on the IR spectra of $YBa_2Cu_3O_x$ and $BiSrCaCu_2O_x$

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Abstract. The assignments for the IR peaks in YBa₂Cu₃O_x were supported by the correlation between the variations of frequencies of IR peaks with lattice constants. The intensities of two peaks at 410 cm⁻¹ (P₄) and 360 cm⁻¹ (P₅) were found to vary with the intensities of peaks at 636 cm⁻¹ (P₁) and 590 cm⁻¹ (P₂), respectively. The IR spectra of BiSrCaCu₂O_x with different oxygen contents were also studied and assignments for the peaks have been suggested.

1. Introduction

IR spectroscopy has been widely used as a powerful technique for the investigation and characterisation of the high- T_c superconductors. It is shown that the IR spectrum of YBa₂Cu₃O_x varies with oxygen content x and reveals changes of frequency and intensity for some peaks [1–10]. In a previous paper [3], we have reported that there are three peaks within the frequency range of 800-400 cm⁻¹ shown as P₁ (620-642 cm⁻¹), P₂ (590 cm⁻¹) and P₃ (540-585 cm⁻¹). Assignments for the peaks were also suggested. At present, however, there is considerable disagreement in the literature for assignments of these peaks [1–3, 8, 9]. For the Bi-Sr-Ca-Cu-O system, there have been to our knowledge no reports concerning the IR spectra of materials with different oxygen contents.

In this work, we studied the IR spectra and lattice constants of $YBa_2Cu_3O_x$ quenched from different temperatures. Some correlation was found between the variations of peak frequencies and lattice constants. The intensity variations of two peaks at 410 cm⁻¹ (P₄) and 360 cm⁻¹ (P₅), which have been reported in literature as one peak, show correlation with that of P₁ and P₂, respectively. We also studied the IR spectra of BiSrCaCu₂O_x with different oxygen contents and assignments for the peaks are discussed.

2. Experimental procedure

The sample preparation and annealing treatment procedures can be found in a previous paper [3]. The lattice constants were measured using x-ray powder diffraction analyses.

Bi-Sr-Ca-Cu-O samples were prepared by solid-state reaction using Bi_2O_3 , SrCO₃, CaCO₃ and CuO with nominal composition 1:1:1:2 ($Bi_1Sr_1Ca_1Cu_2O_3$). The mixed





Figure 1. The IR spectra of $YBa_2Cu_3O_x$ quenched at various temperatures; (*a*), measured by Fourier-transform spectrometer; (*b*), measured by grating infrared spectrophotometer.

powder was ground thoroughly and calcined at 820 °C for 20 h. The calcined powder was then pressed into pellets and sintered at about 882 °C for 48 h followed by annealing at 800 °C for 48 h in air. The sample showed a sharp superconducting transition at about 110 K and eventually reached zero resistance at 95 K. Hydrogen treatment was performed at about 300 °C for different durations in order to induce oxygen vacancies in the structure.

The sample powder was mixed with potassium bromide (KBr) and pressed into pellets for IR measurements. For the frequency range 4900-400 cm⁻¹, a Fourier-transform IR spectrometer (Nicolet 7199C) was used. In order to show the IR peaks with frequencies lower than 400 cm^{-1} , a Perkin–Elmer 577 grating infrared spectro-photometer was also employed.

3. Results

Shown in figure 1 are the IR spectra of $YBa_2Cu_3O_x$ quenched from different temperatures. As no peaks emerge in the frequency range $4900-800 \text{ cm}^{-1}$, we only show the IR spectra within the frequency range $800-300 \text{ cm}^{-1}$. It can be seen that there are five peaks shown as $P_1(620-642 \text{ cm}^{-1})$, $P_2(590 \text{ cm}^{-1})$, $P_3(540-585 \text{ cm}^{-1})$, $P_4(410 \text{ cm}^{-1})$ and $P_5(360 \text{ cm}^{-1})$. The intensity of P_1 increases with increasing quenching temperature and its frequency shifts to higher values; the intensity of P_3 undergoes obvious change but its frequency shifts to lower values; P_2 appears in the spectrum of 700 °C quenched sample and grows rapidly becoming the dominant feature; both P_4 and P_5 increase in



Figure 2. Variation of frequencies ν and lattice constants with quench temperature T_{q} . (a) P₁ peak and lattice constant b. (b) P₃ peak and lattice constant c.

intensity but P_5 appears only in the IR spectra of samples quenched above 700 °C. The intensity variations of P_4 and P_5 show correlations with those of P_1 and P_2 , respectively.

The variations in frequency of P_1 and P_3 and the lattice constants b and c with quenching temperature are shown in figure 2. It shows that the frequency of P_1 increases with decreasing lattice constant b and the frequency of P_3 decreases with increasing lattice constant c.

The IR spectra of BiSrCaCu₂O_x before and after hydrogen treatment at 300 °C for different durations are shown in figure 3. There is only one peak at 610 cm⁻¹ (A₁) in the IR spectrum of BiSrCaCu₂O_x before hydrogen treatment. After hydrogen treatment, three other peaks appear at 475 cm⁻¹ (A₂), 450 cm⁻¹ (A₃) and 370 cm⁻¹ (A₄) in addition to the A₁ peak. The intensities of the four peaks all increase with increasing duration of hydrogen treatment.

4. Discussion

It is generally understood that the frequency of IR peak depends on the bond length of the oscillator. With bond length decreasing, the vibration frequency increases. Figure 2 shows that the frequency of P_1 increases and the lattice constant b decreases with



Figure 3. IR spectra of $BiSrCaCu_2O_x$ before and after hydrogen treatment for different durations; (a), measured by Fourier-transform spectrometer; (b), measured by grating infrared spectrophotometer.



Figure 4. The crystal structure of orthorhombic YBa₂Cu₃O₇.

increasing quench temperature. As the lattice constant b is proportional to the Cu–O bond length in the 1D Cu–O chains (see figure 4), it is reasonable to assign P_1 to the Cu–O of stretching vibration in the 1D Cu–O chains (for the tetragonal phase, it is the Cu–O bond in planes between Ba–Ba layers) [3]. Figure 2 also shows that the frequency of P_3 decreases and lattice constant c increases with increasing quench temperature. This behaviour is very similar to that of 500 cm⁻¹ peak in the Raman spectrum of YBa₂Cu₃O_x, which also shifts to lower frequency values with increasing quench temperature [6, 8, 11–15]. Because the 500 cm⁻¹ Raman peak has been unanimously assigned to the O–Cu–O symmetric stretching vibration (Raman active) along the c axis [1, 6, 8, 11–19], it is

appropriate to attribute P_3 to the O-Cu-O antisymmetric stretching vibration (IR active) along the *c* axis [1, 3]. P_2 has been unanimously assigned to the Cu-O stretching vibration in the 2D Cu-O planes [1-3]. It is noted that the intensity variations of P_4 and P_5 show correlations with those of P_1 and P_2 , respectively. To our knowledge, P_4 and P_5 have been reported in the literature as one peak at 360 cm⁻¹ [2,4]. According to their frequencies and intensity correlations with that of P_1 and P_2 , we assign P_4 and P_5 to the bending modes of Cu-O of the *ab* planes between Ba-Ba layers and Y-Ba layers, respectively.

For BiSrCaCu₂O₂, we have shown in a previous paper [20] that oxygen vacancies can be induced in its structure without the structure being destroyed by hydrogen treatment. These oxygen vacancies have considerable influence on the superconducting characteristics and unit cell volume. As is shown in figure 3, there are four IR peaks. Based on knowledge of the IR spectra of $La_{2-x}Sr_xCuO_4$ and $YBa_2Cu_3O_x$, we tentatively assign the IR peaks of BiSrCaCu₂O_x as follows. The A_1 and A_2 peaks were assigned to the Cu-O and Bi-O stretching vibrations of the Cu-O and Bi-O planes, respectively, because the intensity of in-plane vibration mode is sensitive to the transition from metal to semiconductor [1, 4, 21], these being similar to the P_1 and P_2 peaks in YBa₂Cu₃O_x. Also the frequencies of A_1 and A_2 are consistent with the bond lengths of related oscillators such as Cu–O (1.91 Å) and Bi–O (2.71 Å) [22] (for the structure of the Bi– Sr-Ca-Cu-O system, see reference [22]). The A_4 peak was attributed to the bending modes of Cu–O and Bi–O of the Cu–O and Bi–O planes, these being similar to the P_4 (410 cm^{-1}) and P₅ (360 cm⁻¹) peaks in the IR spectrum of YBa₂Cu₃O_x. A₃ is always present in the semiconductor $BiSrCaCu_2O_r$, but to attribute an assignment to it will require further investigation. The symmetric stretching vibration of oxygen in the Bi-O-Cu-O-Bi chain along its c axis (Raman active) has been found in the Raman spectrum of Bi-Sr-Ca-Cu-O at about 480 cm⁻¹ [22, 23], but we do not observe its IR partner in the IR spectrum, i.e. the antisymmetric stretching vibration of oxygen in Bi-O-Cu-O-Bi (IR active) along its c axis. It may be that it lies hidden in the A_1 peak due to the large linewidth of A_1 .

In summary, we have further studied the IR peaks in $YBa_2Cu_3O_x$. The assignments for P₁, P₂ and P₃ peaks in a previous paper were supported. A new peak (P₄), together with another peak (P₅) were also assigned according to their frequencies and intensity correlations with P₁ and P₂. P₄ and P₅ were assigned to the bending vibrations of the Cu–O bond of the Cu–O planes between Ba–Ba layers and between Y–Ba layers, respectively. We also studied the IR spectrum of BiSrCaCu₂O_x and observed four peaks. A₁ and A₂ were assigned to the stretching vibration of Cu–O and Bi–O of the Cu–O and Bi–O planes, respectively; A₄ (370 cm⁻¹) was attributed to the bending vibrations of Cu– O and Bi–O of the Cu–O and Bi–O planes; A₃ (450 cm⁻¹) needs further investigation.

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